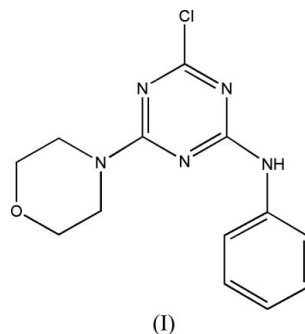
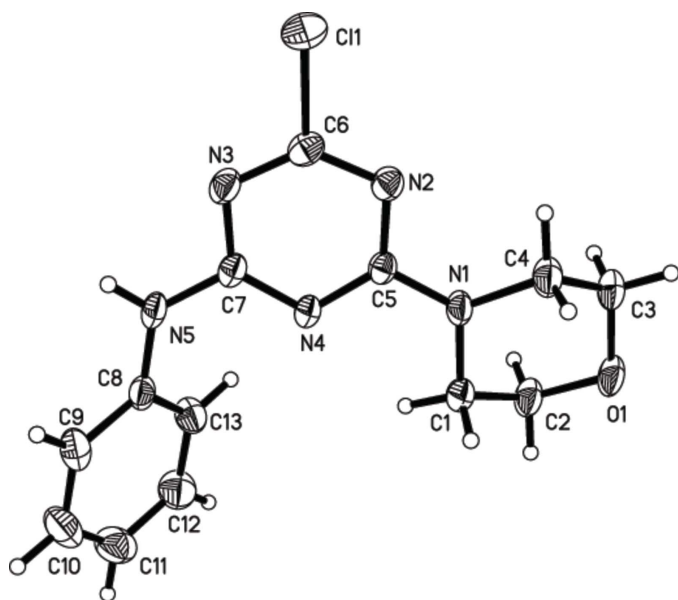


***N*-(4-Chloro-6-morpholino-1,3,5-triazin-2-yl)-  
aniline**Yang Li,<sup>a\*</sup> Li Ding,<sup>b</sup> Tao Zeng,<sup>a</sup>  
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**Key indicators**Single-crystal X-ray study  
*T* = 294 K  
Mean  $\sigma$ (C–C) = 0.003 Å  
*R* factor = 0.039  
*wR* factor = 0.104  
Data-to-parameter ratio = 15.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The title compound, C<sub>13</sub>H<sub>14</sub>ClN<sub>5</sub>O, was synthesized from 2,4-dichloro-6-morpholino-1,3,5-triazine and aniline. In the crystal structure, there are intermolecular N—H···O hydrogen bonds which propagate infinite chains parallel to the *c* axis. The morpholine ring adopts a chair conformation.**Comment**1,3,5-Triazine derivatives are of great interest because of their importance as constituents of active dyes and many hindered amine light stabilizers (Goi, 1960; Degussa, 1968; Manasek & Hrdlovik, 1990). The molecular structure of (I) is shown in Fig. 1. In the crystal structure, molecules are linked by N—H···O hydrogen bonds parallel to the *c* axis. There is also an intermolecular contact which indicates a weak C—H···Cl interaction. The morpholine ring adopts a chair conformation, as expected (see Table 1 and Fig. 2).**Experimental**The title compound was prepared from 2,4-dichloro-6-morpholino-1,3,5-triazine and aniline. 2,4-Dichloro-6-morpholino-1,3,5-triazine was synthesized according to the method described by Dong *et al.* (2005). 2,4-Dichloro-6-morpholino-1,3,5-triazine (23.5 g, 0.1 mol), Na<sub>2</sub>CO<sub>3</sub> (5.51 g, 0.051 mol) and aniline (9.3 g, 0.1 mol) were added to toluene (280 ml), which was stirred at 328–333 K for 5 h. The precipitate was filtered off, the solution was washed with water, and the toluene was evaporated under reduced pressure to give compound (I) in 94.3% yield. Crystals suitable for X-ray diffraction were obtained by slow evaporation of a mixed solution in dichloromethane and ethyl acetate (5:1 *v/v*) (m.p. 438–440 K).**Crystal data**C<sub>13</sub>H<sub>14</sub>ClN<sub>5</sub>O  
*M<sub>r</sub>* = 291.74  
Monoclinic, *P*2<sub>1</sub>/*c*  
*a* = 8.3481 (15) Å  
*b* = 17.667 (3) Å  
*c* = 9.9401 (18) Å  
 $\beta$  = 108.860 (3)°  
*V* = 1387.3 (4) Å<sup>3</sup>  
*Z* = 4*D<sub>x</sub>* = 1.397 Mg m<sup>-3</sup>  
Mo *K*α radiation  
Cell parameters from 2305  
reflections  
 $\theta$  = 2.5–25.7°  
 $\mu$  = 0.28 mm<sup>-1</sup>  
*T* = 294 (2) K  
Block, colourless  
0.28 × 0.22 × 0.20 mm



**Figure 1**  
A perspective view of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

*Data collection*

Bruker SMART CCD area-detector diffractometer	2831 independent reflections
$\varphi$ and $\omega$ scans	1842 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.033$
$T_{\text{min}} = 0.920, T_{\text{max}} = 0.946$	$\theta_{\text{max}} = 26.4^\circ$
7720 measured reflections	$h = -8 \rightarrow 10$
	$k = -16 \rightarrow 22$
	$l = -12 \rightarrow 7$

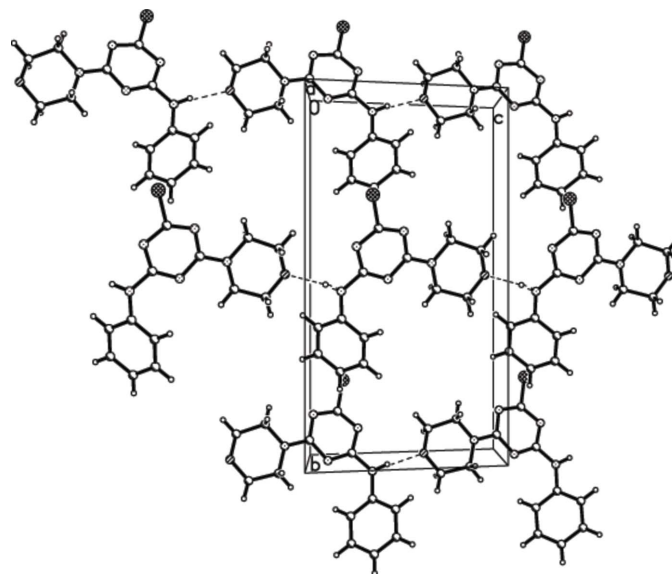
*Refinement*

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0414P)^2 + 0.3828P]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.104$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.01$	$\Delta\rho_{\text{max}} = 0.16 \text{ e } \text{\AA}^{-3}$
2831 reflections	$\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$
185 parameters	
H atoms treated by a mixture of independent and constrained refinement	

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N5-H5 \cdots O1^i$	0.84 (2)	2.08 (2)	2.905 (2)	170 (2)

Symmetry code: (i)  $x, y, z + 1$ .



**Figure 2**  
The crystal structure of (I), viewed down the  $a$  axis. Dashed lines indicate hydrogen-bond interactions.

The N-bound H atom was located in a difference Fourier map and the N–H distance restrained to 0.84 (2)  $\text{\AA}$ . All other H atoms were positioned geometrically and refined using a riding model, with C–H in the range 0.93–0.97  $\text{\AA}$  and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

**References**

Bruker (1997). *SMART, SAINT and SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Degussa, P. (1968). South African Patent ZA 6707089.  
 Dong, C. M., Chen, L. G., Duan, X.-M., Shu, X.-G., Zeng, T. & Yan, X.-L. (2005). *Acta Cryst.* **E61**, o1168–o1169.  
 Goi, M. (1960). *J. Synth. Org. Chem. Jpn.* **18**, 332–336.  
 Manasek, Z. & Hrdlovik, P. (1990). European Patent EP 0377324.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.